Original Article

DESIGN, SYNTHESIS, MOLECULAR DOCKING AND ANTIBACTERIAL EVALUATION OF NOVEL N-(6, 11-DIOXO-DIHYDRO-5H-BENZO [B] CARBAZOL-2YL) BENZAMIDE DERIVATIVES AS POTENT ANTIBACTERIAL AGENTS

P. RAVI CHANDIRAN^a, D. PREMNATH^b, S. VASANTH KUMAR^{a*}

^a Department of Chemistry, School of Science & Humanities, Karunya University, Coimbatore 641114, India. ^b Department of Bioinformatics, School of Biotechnology and Health Sciences, Karunya University, Coimbatore-641 114, India. Email: kumar2359@yahoo.com, ravichandru55@gmail.com

Received: 15 Apr 2014 Revised and Accepted: 14 May 2014

ABSTRACT

Objective: Heterocyclic quinone derivatives are important class of organic compounds both in biological and electrochemical applications. In this article our prime motivation is to develop novel series of heterocyclic quinone derivatives for antibacterial applications. Clinically isolated different Gram-negative and Gram-positive bacterial microorganisms were studied in this article and reported.

Methods: A novel series of N-(6,11-dioxo-dihydro-5H-benzo[b]carbazol-2yl) benzamide derivatives were synthesized by the Michael addition of 1,4-naphthoquinone and *para*-phenylene diamine. The derivatives of compound **1** were synthesized, subjecting it to benzolyation by a variety of acid chlorides. To understand the interaction of binding sites with bacterial protein receptor, the docking study was performed by glide program. *In vitro* antibacterial activity of the synthesized compounds was studied and the MIC value was calculated by the agar dilution method.

Results: The compound **2b** (0.4 μ g/mL) exhibited good antibacterial activity against *staphylococcus aureus* than the standards sparfloxacin (4.87 μ g/mL) and Norfloxacin (39.06 μ g/mL) which was employed.

Conclusion: This investigation identified the potent antibacterial agents and these molecules will be subjected to further studies in our laboratory.

Keywords: Suzuki coupling, Palladium catalyst, 1,4-Naphthoquinone.

INTRODUCTION

In recent years infections caused by bacteria, resistant to multiple antibiotics has been an important problem. Methicillin-resistant Staphylococcus aureus (MRSA) and Vancomycin-resistant Enterococcus (VRE) are the most important infections caused by bacteria which have been found worldwide in hospitals [1]. Thus, the discovery of potent anti-bacterial agents is of great concern inside of multidrug resistance by Gram-positive and gram negative pathogens [2]. Quinones and heterocyclic quinones are large class of compounds with diverse biological activity. They are found to be very cheap and easily available having wide range of biological applications which includes electron transport and oxidative phosphorylation [3]. Heterocyclic quinones are biologically active [4] and heterocyclic aminoquinones have number of successive biological applications including anticancer [5], antibacterial [6-7], fungicidic [7-8], luciferase inhibition [9], antiproliferative [10] and tuberculostatic effects [11]. In addition, the heterocyclic naphthoquinone derivatives exhibit potent properties like electrochemical capacitance [12], electrochemical redox [13], electron mediator [14] and electron transfer [15] in many biological systems. They are also capable molecules of forming complexes with metals [16].In our previous reports [17-18] the novel quinones and heterocyclic quinones were studied for their fluorescent switching properties. In this report our interest is to study the antibacterial activity of novel N- (6,11-dioxodihydro- 5H-benzo [b] carbazol-2yl) benzamide derivatives which is not yet reported in literature, to the best of our knowledge. This present investigation deals with the clinically isolated different gram positive and gram negative bacteria against synthesized compounds and most of the tested compounds act as potent antibacterial agents. To understand the interactions of tested compounds at active sites of protein receptors the molecular docking studies were also preformed and reported in this article.

MATERIALS AND METHODS

Melting points (°C, uncorrected) of all the synthesized compounds were checked in capillary tubes by using a digital melting point apparatus (Labtronics 110, India) and found uncorrected. All the analytical grade chemicals and solvents were purchased from Sigma–Aldrich and Merck, India. Progress and completion of all the reactions were monitored by thin layer chromatography (TLC silica gel 0.25 mm, 60 G F254 and eluting solvents were ethyl acetate: hexane 1:9). All the compounds were characterized by FT-IR spectrometer (IR Prestige-21, Shimadzu, Japan) using KBr pellets, ¹H NMR spectroscopy in DMSO-*d*₆ (500 MHz, Bruker), and ¹³C NMR spectroscopy in DMSO-*d*₆ (125 MHz, Bruker) using tetramethyl silane (TMS) as internal standard. High resolution mass spectra (HRMS-EI) were measured by Electron Impact (EI) method (Jeol GC-Mate 2). Antibacterial studies were carried out by agar dilution method and the MICs were calculated for the tested compounds.

Procedure for synthesis of 2-(4-aminophenylamino)naphthalene-1,4-dione (1)

A mixture of 1,4-naphthoquinone (2 mmol, 0.316 g) and *para*- phenylene diamine (2 mmol, 0.216 g) was added to absolute ethanol (50 mL) and the mixture was refluxed for 10 h. The contents were cooled to room temperature and the reaction mixture was poured into water containing crushed ice and the precipitate formed was collected by vacuum filtration. The product was dried at 50 °C and recrystalized from acetone. Black solid; (0.450 g, 80%); mp > 300 °C; IR (KBr):1120, 1271, 1354, 1436, 1512, 1568, 1670, 3313; ¹H NMR (500 MHz, DMSO-d₆):5.23 (s, 1H), 5.86 (s, 1H), 6.62 (d, 2H, *J* = 8.4 Hz), 7.01 (d, 2H, *J* = 8.0 Hz,), 7.74-7.78 (t, 1H, *J* = 6.4 Hz), 7.82-7.86 (t, 1H, *J* = 6.4 Hz), 7.93 (d, 1H, *J* = 7.6 Hz), 8.96 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆):100.2, 113.9, 114.0, 125.1, 125.2, 125.6, 126.4, 130.4, 132.2, 134.7, 146.9, 181.7, 181.8; HRMS (EI) *m/z:* Calcd for C₁₆H₁₂N₂O₂: 264.2786 Found: 264.2787.

General procedure for synthesis of N-(6,11-dioxo-dihydro-5Hbenzo[b]carbazol 2yl) benzamide derivatives (2a-h)

Substituted acid chlorides (1 mmol) was added to a solution of **1** (0.264 g, 1 mmol) in acetone (100 mL). After refluxing for 30 min, the reaction mixture was filtered and concentrated *in vacuo* to give pure samples of **2a-h** which required no further purification.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)benzamide (2a)

Purple solid; Reaction time 25 minutes (0.345 g, 94%); mp > 300 $^{\circ}$ C; IR (KBr): 1296, 1409, 1514, 1548, 1600, 1672, 3288 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 6.09 (s, 1H), 7.38 (d, 2H, *J* = 9.0 Hz), 7.47-7.56 (m, 4H), 7.59-7.62 (t, 1H, *J* = 6.0 Hz), 7.77-7.81 (t, 1H, *J* = 6.0 Hz), 7.85 (d, 2H, *J* = 6.0 Hz), 7.94-7.97 (t, 1H, *J* = 7.0 Hz), 7.98 (d, 1H, *J* = 7.2 Hz), 8.06 (d, 1H, *J* = 8.0 Hz) 9.23 (s, 1H), 10.36 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 102.1, 121.5, 124.5, 125.7, 126.5, 128.1, 128.8, 129.0, 131.2, 133.0, 135.3, 136.9, 146.7, 165.9, 182.8, 183.5; HRMS (EI) *m/z*: Calcd for C₂₃H₁(h₂O₃: 368.3847 Found: 368.3873.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)-3methylbenzamide (2b)

Purple solid; Reaction time 25 minutes (0.360 g, 94%); mp > 300 $^{\circ}$ C; IR (KBr): 1190, 1234, 1296, 1354, 1408, 1512, 1546, 1598, 1666, 3300, 3365 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 2.41 (s, 3H), 6.09 (s, 1H), 7.37 (d, 2H, *J* = 8.5 Hz), 7.42 (d, 2H, *J* = 8.5 Hz), 7.75-7.80 (m, 4H), 7.84 (t, 2H, *J* = 8.0 Hz), 7.95 (d, 1H, *J* = 7.5 Hz), 8.13 (d, 1H, *J* = 7.5 Hz), 9.23 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 102.1, 121.5, 124.5, 125.2, 126.5, 128.5, 132.6, 135.3, 138.1, 147.0, 166.3, 182.3, 183.1; HRMS (EI) *m/z*: Calcd for C₂₄H₁₈N₂O₃: 382.4113 Found: 382.4112.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)-4methylbenzamide (2c)

Purple solid; Reaction time 25 minutes (0.355g, 94%); mp > 300 °C; IR (KBr): 1118, 1180, 1234, 1355, 1408, 1512, 1595, 1656, 1685, 3271 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 1.18 (s, 3H), 6.02 (s, 1H), 7.31-7.89 (m, 11H), 9.16 (d, 1H, *J* = 8.0 Hz), 10.20 (s, 1H), 12.76 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 20.3, 113.2, 117.0, 122.8, 124.3, 126.8, 132.5, 133.8, 134.8, 135.2, 136.8, 137.7, 138.0, 166.8, 178.2, 181.0; HRMS (EI) *m/z*: Calcd for C₂₄H₁₈N₂O₃: 382.4113 Found: 382.4113.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)-3nitrobenzamide (2d)

Purple solid; Reaction time 25 minutes (0.390 g, 95%); mp > 300 $^{\circ}$ C; IR (KBr): 1120, 1238, 1294, 1352, 1408, 1533, 1616, 1672, 3269 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 6.11 (s, 1H), 7.40 (d, 2H, *J* = 8.0 Hz), 7.78 (d, 2H, *J* = 8.5 Hz), 7.94 (d, 2H, *J* = 8.0 Hz), 8.05 (d, 1H, *J* = 7.5 Hz), 8.33-8.50 (m, 4H), 8.61 (d, 1H, *J* = 8.0 Hz), 9.23 (s, 1H), 10.68 (s, 1H), 13.62 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 102.2, 121.7, 122.8, 124.1, 125.7, 126.5, 127.7, 130.6, 132.9, 133.1, 134.4, 135.3, 136.6, 146.6, 148.2, 163.7, 165.9, 182.0, 182.9; HRMS (EI) *m/z*: Calcd for C₂₃H₁₅N₃O₅: 413.3823 Found: 413.3822.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)-4-nitrobenzamide (2e)

Purple solid; Reaction time 25 minutes (0.385 g, 93%); mp > 300 $^{\circ}$ C; IR (KBr): 1109, 1296, 1350, 1413, 1535, 1602, 1678, 3263 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) & 6.11 (s, 1H), 7.41 (d, 2H, *J* = 8.5 Hz), 7.79 (t, 1H, *J* = 8.2 Hz), 7.85 (d, 2H, *J* = 7.5 Hz), 7.95 (d, 2H, *J* = 6.8 Hz), 8.07 (d, 1H, *J* = 7.0 Hz), 8.16 (d, 2H, *J* = 8.0 Hz), 8.32 (d, 2H, *J* = 8.5 Hz), 8.38 (t, 1H, *J* = 7.5 Hz), 9.25 (s, 1H), 10.67 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) & 121.7, 124.0, 124.2, 124.5, 125.7, 129.7, 131.1, 136.8, 150.5, 160.5, 166.2, 180.0, 181.3; HRMS (EI) *m/z:* Calcd for C₂₃H₁₅N₂O₅: 413.3823 Found: 413.3822.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)-3,5-dinitrobenzamide (2f)

Purple solid; Reaction time 25 minutes (0.435 g, 95%); mp > 300 °C; IR (KBr): 1080, 1122, 1159, 1242, 1294, 1344, 1409, 1516, 1543, 1668, 3078, 3267 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 6.13 (s, 1H), 7.14 (t, 1H, *J* = 8.0 Hz), 7.23 (t, 1H, *J* = 7.5 Hz), 7.34 (d, 1H, *J* = 7.5 Hz), 7.44 (d, 2H, *J* = 8.0 Hz), 7.78 (d, 1H, *J* = 8.5 Hz), 7.86 (d, 2H, *J* = 8.0 Hz), 8.08 (s, 1H), 9.26 (s, 1H), 10.96 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 102.3, 121.6, 122.5, 124.5, 125.7, 126.5, 128.5, 129.3, 130.8, 133.1, 135.9, 137.8, 148.2, 163.5, 182.0, 182.9; HRMS (EI) *m/z*: Calcd for C₂₃H₁₄N₄O₇: 458.3798 Found: 458.3798.

N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)acetamide (2g)

Purple solid; Reaction time 25 minutes (0.290 g, 95%); mp > 300 $^{\circ}$ C; IR (KBr): 1122, 1269, 1294, 1357, 1409, 1523, 1566, 1606, 1683, 3194, 3309 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 2.06 (s, 3H), 6.03 (s, 1H), 7.30 (d, 2H, *J* = 8.5 Hz), 7.63 (d, 2H, *J* = 9.0 Hz), 7.77 (t, 1H, *J* = 7.5 Hz), 7.84 (t, 1H, *J* = 7.5 Hz), 7.94 (d, 1H, *J* = 7.5 Hz), 8.05 (d, 1H, *J* = 7.5 Hz), 9.17 (s, 1H), 9.26 (s, 1H), 10.04 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 24.7, 101.9, 120.1, 124.7, 125.7, 126.5, 130.9, 133.0, 135.3, 171.7, 183.0; HRMS (EI) *m/z*: Calcd for C₁₈H₁₄N₂O₃: 306.3153 Found: 306.3153.

2-chloro-N-(4-(1,4-dioxo-1,4-dihydronaphthalene-2ylamino)phenyl)acetamide (2h)

Purple solid; Reaction time 25 minutes (0.320 g, 94%); mp > 300 $^{\circ}$ C; IR (KBr): 989, 1124, 1294, 1359, 1409, 1521, 1564, 1597, 1676, 3076, 3190 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 4.26 (s, 2H), 6.06 (s, 1H), 7.35 (d, 2H, *J* = 8.5 Hz), 7.66 (d, 2H, *J* = 8.5 Hz), 7.76 (t, 1H, *J* = 7.5 Hz), 7.84 (t, 1H, *J* = 7.5 Hz), 7.93 (d, 1H, *J* = 8.0 Hz), 8.05 (d, 1H, *J* = 7.5 Hz), 9.20 (s, 1H), 10.40 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 44.01, 102.1, 120.6, 124.7, 125.7, 126.5, 126.5, 130.8, 133.0, 133.1, 134.1, 135.3, 136.1, 146.7, 165.0, 182.0, 182.8; HRMS (EI) *m/z*: Calcd for C₁₈H₁₃ClN₂O₃: 340.7604 Found: 340.7604.

General Procedure for synthesis of N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)benzamide derivatives (3a-h)

Mixture of **2a-h** (0.5 mmol) in glacial acetic acid (60 mL) and palladium (II) acetate (0.112 g, 0.5 mmol) were refluxed for 2 h and the reaction mixture was cooled at room temperature and poured into ice cold water. The precipitate was filtered, dried at 60 $^{\circ}$ C and crystallized from acetone to give **3a-h**.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)benzamide (3a)

Yellow solid; Reaction time 2 hours (0.265 g, 73%); mp > 300 $^{\circ}$ C; IR (KBr): 1010, 1240, 1271, 1375, 1481, 1587, 1647, 3277 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 7.45-7.61 (m, 5H), 7.80-7.89 (m, 4H), 8.02 (d, 2H, *J* = 7.0 Hz), 8.75 (s, 1H), 10.43 (s, 1H), 13.09 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 113.4, 114.2, 122.2, 124.6, 126.4, 128.1, 132.0, 133.1, 134.6, 135.4, 136.3, 137.9, 165.9, 177.8, 180.6; HRMS (EI) *m/z*: Calcd for C₂₃H₁₄N₂O₃: 366.3688 Found: 366.3687.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)-3methylbenzamide (3b)

Yellow solid; Reaction time 2 hours (0.280 g, 74%); mp > 300 °C; IR (KBr): 1012, 1238, 1273, 1321, 1377, 1431, 1481, 1535, 1591, 1645, 1668, 2920, 3277 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) & 2.43 (s, 3H), 7.43 (d, 2H, *J* = 7.0 Hz), 7.59 (d, 1H, *J* = 8.0 Hz), 7.81 (d, 1H, *J* = 7.5 Hz), 7.83 (d, 1H, *J* = 6.0 Hz), 7.89 (t, 1H, *J* = 8.0 Hz), 8.10 (d, 1H, *J* = 7.0 Hz), 7.59 (d, 1H, *J* = 8.0 Hz), 8.10 (d, 1H, *J* = 7.0 Hz), 7.59 (d, 1H, *J* = 8.0 Hz), 8.10 (d, 1H, *J* = 7.0 Hz), 7.59 (d, 1H, *J* = 8.0 Hz), 8.10 (d, 1H, *J* = 7.0 Hz), 7.59 (d, 1H, *J* = 8.0 Hz), 8.13 (d, 1H, *J* = 8.0 Hz), 8.74 (s, 1H), 10.37 (s, 1H), 13.08 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) & 21.4, 113.3, 114.2, 117.8, 122.2, 124.6, 125.3, 126.4, 128.6, 132.5, 133.1, 134.6, 135.4, 136.3, 137.9, 138.1, 166.0, 177.8, 180.6; HRMS (EI) *m/z*: Calcd for C₂₄ H₁₆N₂O₃: 380.3954 Found: 380.3954.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)-4methylbenzamide (3c)

Yellow solid; Reaction time 2 hours (0.278g, 74%); mp > 300 °C; IR (KBr): 1010, 1273, 1323, 1483, 1527, 1589, 1641, 1668, 2358, 2848, 2918, 3267 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 1.23 (s, 3H), 7.35 (d, 2H, *J* = 8.0 Hz), 7.57 (d, 1H, *J* = 8.5 Hz) 7.81-7.89 (m, 3H), 7.94 (d, 2H, *J* = 8.0 Hz), 8.10 (d, 1H, *J* = 7.5 Hz), 8.15 (d, 1H, *J* = 7.5 Hz), 8.74 (s, 1H), 10.33 (s, 1H), 13.08 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 21.3, 114.7, 115.3, 1121.3, 124.7, 128.8, 133.2, 134.0, 135.0, 136.3, 137.5, 138.8, 166.1, 177.0, 178.3; HRMS (EI) *m/z*: Calcd for C₂₄H₁6N₂O₃: 380.3954 Found: 380.3953.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)-3nitrobenzamide (3d)

Yellow solid; Reaction time 2 hours (0.300 g, 72%); mp > 300 $^{\circ}$ C; IR (KBr): 1010, 1240, 1271, 1327, 1485, 1527, 1591, 1668, 2918, 3205 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 7.61 (d, 1H, *J* = 6.8 Hz), 7.69 (d,

1H, J = 7.5 Hz), 7.83-7.99 (m, 7H), 8.74 (s, 1H), 8.87 (s, 1H), 10.75 (s, 1H), 13.11 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 105.3, 110.4, 113.6, 118.1, 122.1, 126.5, 130.6, 134.6, 135.7, 136.7, 148.2, 163.7, 177.8, 178.3; HRMS (EI) m/z: Calcd for C₂₃H₁₃N₃O₅: 411.3664 Found: 411.3664.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)-4nitrobenzamide (3e)

Yellow solid; Reaction time 2 hours (0.305 g, 74%); mp > 300 $^{\circ}$ C; IR (KBr): 1010, 1273, 1346, 1523, 1595, 1674, 2918, 3263 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 7.75 (d, 2H, *J* = 8.0 Hz), 7.73 (d, 2H, *J* = 8.0 Hz), 8.05-8.43 (m, 7H), 10.74 (s, 1H), 13.13 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 99.8, 124.0, 127.4, 129.7, 141.2, 153.8, 163.2, 180.1, 181.7; HRMS (EI) *m/z*: Calcd for C₂₃H₁₃N₃O₅: 411.3664 Found: 411.3664.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)-3,5dinitrobenzamide (3f)

Yellow solid; Reaction time 2 hours (0.340 g, 75%); mp > 300 $^{\circ}$ C; IR (KBr): 1016, 1155, 1246, 1330, 1489, 1541, 1587, 1666, 2918, 3084, 3203, 3390 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 7.18 (d, 1H, *J* = 5.8 Hz), 7.36 (d, 1H, *J* = 6.2 Hz), 7.76-8.02 (m, 7H), 8.72 (s, 1H), 11.01 (s, 1H), 13.15 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 100.18, 102.7, 128.5, 134.8, 138.4, 148.6, 169.5, 180.2, 181.7; HRMS (EI) *m/z*): Calcd for C₂₃H₁₂N₄O₇: 456.3639 Found: 456.3639.

N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)acetamide (3g)

Yellow solid; Reaction time 2 hours (0.230 g, 75%); mp > 300 $^{\circ}$ C; IR (KBr): 1008, 1240, 1273, 1369, 1516, 1589, 1647, 2926, 3442 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 1.23 (s, 3H), 7.51 (d, 1H, *J* = 8.5 Hz), 7.64 (d, 1H, *J* = 8.0 Hz), 7.71 (s, 1H), 7.80-7.90 (m, 4H), 10.09 (s, 1H), 13.02 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 24.4, 114.4, 126.5, 133.2, 134.6, 168.9, 180.7, 183.0; HRMS (EI) *m/z*: Calcd for C₁₈H₁₂N₂O₃: 304.2994 Found: 304.2994.

2-chloro-N-(6,11-dioxo-6,11-dihydro-5H-benzo[b]carbazol-2-yl)acetamide (3h)

Yellow solid; Reaction time 2 hours (0.250 g, 74%); mp > 300 $^{\circ}$ C; IR (KBr): 1006, 1273, 1375, 1531, 1587, 1672, 3275, 3739 cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ : 2.09 (s, 2H), 7.51 (d, 1H, *J* = 6.5 Hz), 7.63 (d, 1H, *J* = 7.5 Hz), 7.80-7.90 (m, 2H), 8.09-8.16 (m, 2H), 10.08 (s, 1H), 10.47 (s, 1H), 13.08 (s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ : 69.0, 101.3, 102.6, 102.7, 103.1, 104.7, 105.8, 110.3, 113.8, 114.1, 118.2, 121.3, 126.8, 130.2, 132.0, 135.8, 137.8, 141.2, 163.2, 171.4, 178.2; HRMS (EI) *m/z*: Calcd for C₁₈H₁₁ClN₂O₃: 338.7445 Found: 338.7445.

Molecular docking studies

To understand the interaction of all the synthesized molecules (1, **2a-h**, **3a-h**) with *Bacillus substillis*, the crystal structure of YmaH from Bacillus subtillis [21] were downloaded from protein data bank and the molecular docking studies were performed using the GLIDE program [22] (version 8.5, Schrodinger, LLC, New York, 2010). To analyze the docking results and execute the protocol, the maestro user interface (version 8.5, Schrodinger, LLC, New York, 2010) was employed and the validation of protocol was evaluated by redocking. YmaH (PDB ID: 3HSB) were selected for docking studies as a reference sample and was prepared for docking through protein preparation wizard. Structures of **1**, **2a-h**, **3a-h** were sketched using ACD/chemsketch (Freeware version). GLIDE grid generation wizard has been used to define the docking space. Docking was performed using XP (Extra Precision mode) docking protocol.

In vitro Antibacterial activity

All the synthesized compounds were studied for their antibacterial activity against clinically isolated two Gram-positive bacteria (*Bacillus subtillisn and Klebsiella Pneumoniae*) and five Gramnegative bacilli (*Staphylcoccous aureus*, *Escherichia coli*, *Proteus vulgaris*, *Salmonella typhi*, *Pseudomonas aureus*) using conventional agar-dilution method [23-24]. The minimum inhibitory concentrations (MICs) values were calculated by comparison to Sparfloxacin and Norfloxacin as the reference bacterial drugs and they are presented in Table 1. All the cultures were prepared by Muller Hinton agar and the turbidity of all the bacterial cultures was adjusted to 0.5 McFarland Standard by preparing bacterial suspension of 3-5 well-isolated colonies of the same morphological type selected from an agar plate culture. The cultures were further diluted 1000-fold to get an inoculums size of 1.5 X 105 CFU/The synthesized compounds and standard bacterial drugs (50 mg) were dissolved in Dimethyl formamide (DMF) (0.5 mL) and the solution was diluted with water (4.5 mL) to get a stock solution of 10.000 mg/L of each compound. Further progressive double dilution with Muller-Hinton broth was performed to obtain the required concentrations of 2500-0.7 µg/mL [25]. To ensure that the solvent had no effect on the bacterial growth, a control test was performed with a test medium supplemented with DMF at the same dilutions as used in the experiment.

In each micro well inoculated with 75 μL of the serial dilutions, 75 μL of the bacterial suspension was added in a series of 12 micro wells. Incubation of the cultures overnight at 37 $^{\rm Q}C$ was done and the growth measured. The MICs of the test compounds and the standard control drugs are tabulated in Table 1.

RESULTS AND DISCUSSION

Chemistry

In our previous report, simple "off-on-off" chemical and switches electrochemical fluorescent were successfully demonstrated [18]. The compound (1) was synthesized by the nucleophilic amino addition reaction of equal mole concentrations of 1,4-naphthoquinone and para-phenylene diamine in the presence of absolute ethanol. The mixture was refluxed for 10 h and the yield obtained was 85%. In a previous report [19] the same compound (1) was synthesized by amino substitution in the presence of absolute ethanol at room temperature. The product was obtained after 96 h of stirring and the percentage of yield compound **1** was not given. In another report [20] the same reaction was carried out in the presence of water-glacial acetic acid mixture under reflux condition. The yield reported was between 49-62%. In one of our previous study we reported [17] the nucleophilic amino addition and substitution reactions with 1,4-quinone moiety in the presence of ethanol, water and solvent free microwave system. Based on the above literature we attempted the amino substitution reaction with 1,4-naphthoquinone in the presence of water as solvent and the reaction was not successful. A number of inseparable compounds were formed and the yield was low (25%) compared to ethanol mediated amino addition. In the second step, the intra molecular carbon-carbon bond linkage was carried out by palladium (II) acetate in the presence of glacial acetic acid and the product yield was 72-75%. The percentage of yield for compounds (3a-h) is comparatively very good. In the reported work [19] no information about the product yield was given and in another report [20] the yields were 18-62% for the palladium catalyzed reaction.

Biology

In vitro antibacterial studies of quinone derivatives

All the synthesized compounds were tested against two Grampositive and five Gram-negative bacteria's. All the compounds (1, 2a-h, 3a-h) exhibited good bacterial activity against gram positive bacteria of Klebsiella Pneumoniae than the standard drugs used (Sparfloxacin and Norfloxacin). Compound 3f exhibits good activity against most of the gram positive and gram negative microorganisms due to the presence of the two nitro groups at the third and fifth position of the aromatic system of the benzoyl unit. Compound 2f exhibits better activity against Escherichia coli (98 µg/mL) than Sparfloxacin (156.3 µg/mL) and Norfloxacin (625 μ g/mL). Compound 1 (1.2 μ g/mL) exhibits better activity against Proteus vulgaris than the standard drug Sparfloxacin (4.8 µg/mL). Compound **3c** (227 μ g/mL) exhibits better bacterial activity against Salmonella typhi than Sparfloxacin (2500 µg/mL) and Norfloxacin (627 µg/mL). Compound 3a (32 µg/mL), 3b (27 µg/mL), 3c (23 µg/mL) exhibits better antibacterial activity against Pseudomonas aureus than the standards of Sparfloxacin (156.3 µg/mL) and

Norfloxacin (39.06 μ g/mL). Compounds **2b** (0.4 μ g/mL) and **2f** (0.4 μ g/mL) exhibited very good antibacterial activity due to the presence of methyl and nitro functional groups, among all the molecules synthesized against *Staphylococcus aureus* than Sparfloxacin (4.87 μ g/mL) and Norfloxacin (39.06 μ g/mL). Standard

drug Norfloxacin could not exhibit any activity against *Bacillus subtillis* and *Proteus vulgaris* microorganisms. Compound **2g** did not exhibit any inhibition against *Bacillus subtillis, Escherichia coli, Proteus vulgaris* and compound **3d** did not exhibited any inhibition against *Escherichia coli.*



2h, 3h = Methylene chloride



Table 1: *In vitro* antibacterial activity of synthesized compounds against Gram-positive and Gram-negative bacteria (MICs in µg/mL) Values were the means of three replicates ± SD.

Compounds	MIC (µg/mL)						
	B. subtilis	S. aureus	E. coli	P.vulgaris	S. typhi	P.aureus	K. Pneumoninae
1	33.9 ± 0.52	29 ± 0.25	621 ± 0.29	1.2 ± 0.10	431 ± 0.52	113 ± 0.88	214 ± 0.81
2a	52.8 ± 0.23	17 ± 0.28	98 ± 0.19	12 ± 0.23	521 ± 0.12	63.5 ± 0.85	387 ± 0.65
2b	52 ± 0.83	0.4 ± 0.12	216 ± 0.23	21 ± 0.59	523 ± 0.36	145 ± 0.63	287 ± 0.33
2c	26.1 ± 0.22	2.6 ± 0.23	214 ± 0.92	18 ± 0.56	432 ± 0.89	136 ± 0.92	298 ± 0.92
2d	29.3 ± 0.12	523 ± 0.89	214 ± 0.89	453 ± 0.81	752 ± 0.53	57 ± 0.93	278 ± 0.81
2e	31.7 ± 0.33	28.6 ± 0.32	213 ± 0.22	18 ± 0.72	654 ± 0.72	122 ± 0.33	342 ± 0.93
2f	52 ± 0.11	0.4 ± 0.18	216 ± 0.36	21 ± 0.21	523 ± 0.36	145 ± 1.02	287 ± 0.33
2g	*	36 ± 0.59	*	*	523 ± 0.72	54 ± 0.99	521 ± 0.92
2h	31.2 ± 0.82	41 ± 0.93	523 ± 0.82	19 ± 0.92	346 ± 0.36	132 ± 0.33	647 ± 0.25
3a	524 ± 0.36	19 ± 0.36	348 ± 0.52	523 ± 1.23	892 ± 0.80	32 ± 0.82	432 ± 1.23
3b	457 ± 0.87	14.53 ± 0.91	321.5 ± 0.23	568 ± 1.32	1052 ± 1.28	27 ± 0.91	654 ± 1.02
3c	8.8 ± 0.12	12.4 ± 0.25	391 ± 0.23	47 ± 1.09	227 ± 0.82	23 ± 1.22	897 ± 0.92
3d	21.3 ± 022	25 ± 0.82	*	13 ± 0.93	324 ± 0.69	91 ± 1.39	826 ± 0.24
3e	53 ± 0.33	17 ± 0.36	356 ± 0.82	27 ± 0.56	500 ± 0.36	142 ± 0.93	973 ± 0.78
3f	6 ± 0.09	23.5 ± 0.25	326 ± 1.09	523 ± 0.23	563 ± 0.85	14 ± 0.23	825 ± 0.82
3g	94 ± 0.17	6.8 ± 0.83	489 ± 0.23	32 ± 0.89	765 ± 0.22	124 ± 0.89	956 ± 0.71
3h	83.1 ± 0.41	4.5 ± 0.89	567 ± 0.29	321 ± 0.56	523 ± 1.09	83 ± 0.99	2000 ± 0.23
Sparfloxacin ^a	9.76 ± 0.52	4.87 ± 0.25	156.3 ± 0.89	4.8 ± 0.27	2500 ± 0.99	156 ± 0.39	2500 ± 1.22
Norfloxacin ^a	*	39.06 ± 0.23	625 ± 1.20	*	627 ± 0.52	39.06 ± 1.23	<1.2 ± 0.92

* No inhibition observed a Standard antibacterial drugs Lower MIC values indicates that higher antimicrobial activity Bold letters indicates better activity against microorganisms

Molecular docking studies of quinone derivatives

To understand the interaction of bacterial protein receptor with synthesized molecules (**1**, **2a-h**, **3a-h**) the crystal structure of YmaH from Bacillus subtillis was downloaded from protein data bank and studied with the glide program. The entire glide, E model scores and hydrogen bonds interactions are compared with the MICs of Bacillus subtillis for the tested compounds a are presented in table 2. The use of

glide and E model scores for ranking the different derivatives within a series is always not dependable. The molecular docking and *in vitro* antibacterial study results show that the glide scores and MIC values of the synthesized compounds do not have any correlation. The glide scores are mainly used to identify the active and inactive compounds. In addition, glide is primarily concerned with generating an accurate pose for each ligand and enrichment (the separation of actives from inactives) (See figures 1-4) [22, 26].

Compounds	Glide score	E model score (kcal/mol)	Molecular docking		
			No. of Hydrogen bonds interactions	MICs of BS (µg/mL)	
1	-4.92	-34.71	2 (ASP 269, HIS 180)	33.9	
2a	-5.33	-57.97	1 (LEU 142)	52.8	
2b	-4.47	-60.14	2 (HIE 268, LYS 179)	52	
2c	-5.86	-63.79	1 (GLN 63)	26.1	
2d	-5.05	-63.29	1 (ASP 269)	29.3	
2e	-5.12	-64.62	2 (ASP 269, LYS 179)	31.7	
2f	-3.86	-75.11	Hydrophophic interaction	52	
2g	-5.16	-49.42	1 (ASP 269)	*	
2h	-5.78	-60.78	2 (GLN 63, GLN 208)	31.2	
3a	-3.26	-63.54	1 (ASN 273)	524	
3b	-3.96	-70.62	1 (ASN 273)	457	
3c	-3.96	-66.01	1 (ASN 273)	8.8	
3d	-3.16	-72.76	1 (ASN 273)	21.3	
3e	-1.01	-69.72	1 (ASN 273)	53	
3f	-4.24	-72.39	2 (ASP 274, TRP 58)	6	
3g	-5.77	-44.69	1 (GLN 208)	94	
3h	-4.87	-52.74	1 (GLN 63)	83.1	
Sparfloxacin	-	-	-	9.76	
Norfloxacin	-	-	-	*	

Table 2: Molecular docking studies of ten analogues taken for study with Bacillus subtillis (PDB ID: 3HSB)

-Docking studies not carried out * No inhibition observed Bold letters indicates better activity glide score against Bacillus subtillis (BS)



Fig. 1: Docking model structure of compound 2c respectively into the YmaH binding pocket.



Fig. 2: Docking model structure of compound 2h respectively into the YmaH binding pocket.



Fig. 3: Docking model structure of compound 3f respectively into the YmaH binding pocket.



Fig. 4: Docking model structure of compound 3g respectively into the YmaH binding pocket.

CONCLUSION

In summary, a new series of novel N-(6,11-dioxo-dihydro-5Hbenzo[b]carbazol-2yl) benzamide derivatives were synthesized and characterized by FT-IR, 1H, 13C NMR and high resolution mass (HRMS-EI) spectral analyses. All the molecules were studied for their interactions with YmaH by molecular docking protocol. Among the tested molecules, compound 2c exhibited a good glide score value of 5.86 with e model value of -63.79. In vitro antibacterial activity of the tested compounds shows improved activity against all the microorganisms used. In particular compound 2f exhibits marked activity against two microorganisms. Compound 2b and 2f (0.4 μ g/mL) exhibits good activity against Staphylococcus aureus than Sparfloxacin (4.87 μ g/mL) and Norfloxacin used (39.06 μ g/mL).

CONFLICT OF INTEREST

Declared None

ACKNOWLEDGEMENTS

The authors thank the Management and the authorities of Karunya University, Coimbatore, for their kind support, constant encouragement and also for providing KSJF fellowship to PR. Our thanks are also extended to SAIF, IIT, Madras, India for NMR and HR-Mass spectral analysis.

REFERENCES

- 1. Yang Z, Kitano Y, Chiba K, Shibata N, Kurokawa H, Doi Y, et al. Synthesis of variously oxidized abietane diterpenes and their antibacterial activities against MRSA and VRE. Bioorganic & medicinal chemistry 2001;9(2):347-56.
- Shin D-Y, Kim SN, Chae J-H, Hyun S-S, Seo S-Y, Lee Y-S, et al. Syntheses and anti-MRSA activities of the C3 analogs of mansonone F, a potent anti-bacterial sesquiterpenoid: insights into its structural requirements for anti-MRSA activity. Bioorganic & medicinal chemistry letters 2004;14(17):4519-23.
- Lee H-J, Kim JS, Suh M-E, Park HJ, Lee SK, Rhee H-K, et al. Synthesis and cytotoxicity evaluation of substituted pyridazino [4,5-b]phenazine-5,12-diones and tri/tetra-azabenzofluorene-5,6-diones. European journal of medicinal chemistry 2007; 42 (2):168-74.
- Chung H-J, Jung O-J, Chae MJ, Hong S-Y, Chung K-H, Lee SK, et al. Synthesis and biological evaluation of quinoxaline-5,8-diones that inhibit vascular smooth muscle cell proliferation. Bioorganic & medicinal chemistry letters 2005;15(14):3380-4.
- Lee H-J, Suh M-E, Lee C-O. Synthesis and cytotoxicity evaluation of 2-amino- and 2-hidroxy-3-ethoxycarbonyl-N-substitutedbenzo[f]indole-4,9-dione derivatives. Bioorganic & medicinal chemistry 2003;11(7):1511-19.
- Tandon VK, Maurya HK, Verma MK, Kumar R, Shukla PK. 'On water' assisted synthesis and biological evaluation of nitrogen and sulfur containing hetero-1,4-naphthoquinones as potent antifungal and antibacterial agents. European journal of medicinal chemistry 2010;45(6):2418-26.
- Bhaskar G, Arun Y, Balachandran C, Saikumar C, Perumal PT. Synthesis of novel spirooxindole derivatives by one pot multicomponent reaction and their antimicrobial activity. European journal of medicinal chemistry 2012; 51:79-91.
- Tandon VK, Maurya KK, Tripathi A, ShivaKeshava GB, Shukla PK, Srivastava P et al. 2,3-Disubstituted-1,4-naphthoquinones, 12Hbenzo[b]phenothiazine-6,11-diones and related compounds: Synthesis and Biological evaluation as potential antiproliferative and antifungal agents. Eur J Med Chem 2009; 44: 1086-1092.
- Bedford R, LePage D, Hoffmann R, Kennedy S, Gutschenritter T, Bull L et al. Luciferase inhibition by a novel naphthoquinone. J Photochem PhotoBiol b Biology 2012; 107: 55–64.

- Tandon VK, Chhor RB, Singh RV, Rai S, Yadav DB. Design, synthesis and evaluation of novel 1,4-naphthoquinone derivatives as antifungal and anticancer agents. Bioorg Med Chem Lett 2004; 14: 1079-1083.
- Osman SAA, Abdalla AA, Alaib MO. Synthesis of sulfanilamidonaphthoquinones as potential anti-tuberculous agents. J Pharm Sci 1983; 72: 68-71.
- Leitner KW, Gollas B, Winter M, Besenhard JO. Combination of redox capacity and double layer capacitance in composite electrodes through immobilization of an organic redox couple on carbon black. Electrochim Acta 2004; 50: 199-204.
- 13. Yucel B, Sanli B, Soylemez H, Yilmaz I. "Synthesis and electro-spectro electrochemistry of ferrocenyl naphtha quinones. Tetrahedron 2011; 67: 1406-1421.
- 14. Manisankar P, Pushpalatha AM, Vasanthkumar S, Gomathi A, Viswanathan S. "Riboflavin as an electron mediator catalyzing the electrochemical reduction of dioxygen with 1,4naphthoquinones. J Electroanal Chem 2004; 571: 43–50.
- Abhayawardhana AD, Sutherland TC. "Heterogeneous protoncoupled electron transfer of a hydroxy-anthraquinone selfassembled monolayer. J Electroanal Chem 2011; 653: 50–55.
- Afrasiabi Z, Sinn E, Chen J, Ma Y, Rheingold AL, Zakharov LN et al."Appended 1,2-naphthoquinones as anticancer agents 1: synthesis, structural, spectral and antitumor activities of orthonaphthaquinone thiosemicarbazone and its transition metal complexes. Inorg Chim Acta 2004; 357: 271–278.
- Ravichandiran P, Kannan R, Ramasubbu A, Muthusubramanian S, Samuel VK. "Green synthesis of 1,4-quinone derivatives and evaluation of their fluorescent and electrochemical properties. J Saudi Chem Soci 2012; http:// dx.doi.org/ 10.1016/ j.jscs. 2012.09.011.
- Ravichandiran P, Santhoshkumar P, Vasanthkumar S. Green synthesis of 1,4-quinone derivatives and evaluation of their fluorescent and electrochemical properties J Saudi Chem Soci 2013; http://dx.doi.org/10.1016/j.jscs.2013.08.002.
- Illos RA, Shamir D, Shimon LJW, Zilbermann I, Bittner S. "N-Dansyl-carbazoloquinone; a chemical and electrochemical fluorescent switch. Tetrahedron Lett 2006; 47: 5543–5546.
- 20. Bernardo PH, Chai CLL, Guen ML, Smith GD, Waring P. "Structureactivity delineation of quinones related to the biologically active Calothrixin B. Bioorg Med Chem Lett 2007; 17: 82-85.
- Someya T, Baba S, Fujimoto M, Kawai G, Kumasaka T, Nakamura K. Crystal structure of YmaH (Hfq) from Bacillus subtilis in complex with an RNA aptamer. Nucleic Acids Res 2012; 40: 1856-1867.
- Friesner RAO, Murphy RB, Repasky MP, Frye LL, Greenwood JR, Halgren TA et al. Extra Precision GlideDocking and Scoring Incorporating a Model of Hydrophobic Enclosure for Protein–Ligand Complexes. J Med Chem 2006; 49: 6177–6196.
- Baron EJ, Finegold SM. In Bailey and Scott's Diagnostic Microbiology, 8th ed.; C.V. Mosby: St. Louis 1990; 184–188.
- 24. Massah AR, Adibi H, Khodarahmi R, Abiri R, Majnooni MB, Shahidi S et al. Synthesis, in vitro antibacterial and carbonic anhydrase II inhibitory activities of N-acylsulfonamides using silica sulfuric acid as an efficient catalyst under both solventfree and heterogeneous conditions. Bioorg Med Chem 2008; 16: 5465- 5472.
- 25. Andrews JMJ. Determination of minimum inhibitory concentrations. Antimicrob Chemother 2001; 48: 5-16.
- 26. Source: http://www.schrodinger.com/kb/144